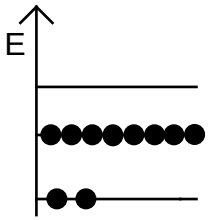
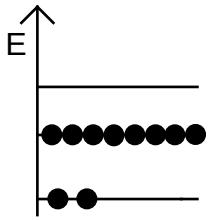
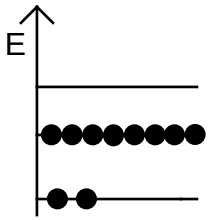
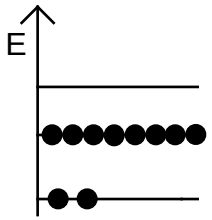
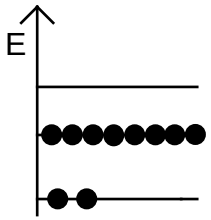
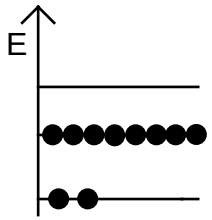
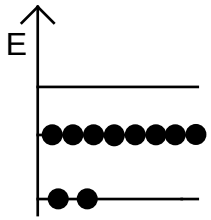
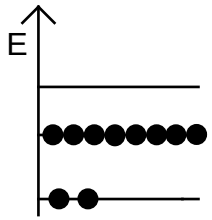
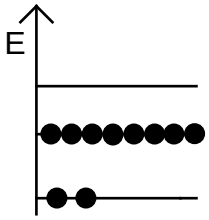
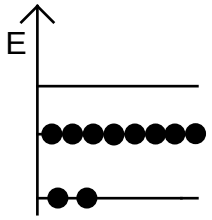
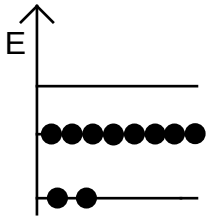
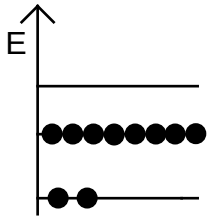
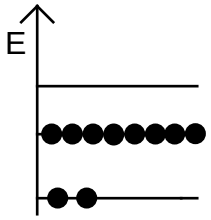
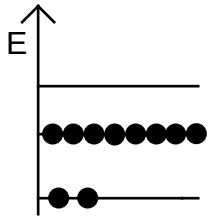
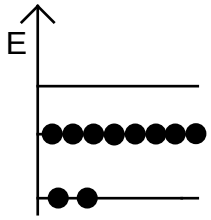
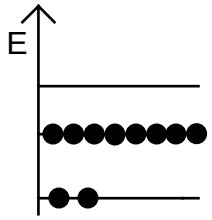
Arbeitsblatt	Zusammenhang zwischen der Ionenladung und der Stellung der Hauptgruppenelemente im PSE	Klasse 8

I. HG	II. HG	III. HG	IV. HG	V. HG	VI. HG	VII. HG	VIII. HG
 $_{11}$Na	 $_{12}$Mg	 $_{13}$Al	 $_{14}$Si	 $_{15}$P	 $_{16}$S	 $_{17}$Cl	 $_{18}$Ar
 $_{11}$Na ☐	 $_{12}$Mg ☐	 $_{13}$Al ☐	 $_{14}$Si ☐	 $_{15}$P ☐	 $_{16}$S ☐	 $_{17}$Cl ☐	 $_{18}$Ar ☐

Bildung von .....ionen durch	Bildung meist von .....ionen	Bildung von .....ionen durch	..... Bildung von Ionen
für Elemente der 3. Periode: Erreichen des Edelgaszustandes des Elements		*für Elemente der 3. Periode:* Erreichen des Edelgaszustandes des Elements	Atome sind .....
Ladung der Ionen entspricht		Ladung der Ionen:	Edelgase sind .....